Chemo-thermo-geomechanical implications for monitoring CCS (and geothermal) reservoirs

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We Need Abatement, CCS, & Negative Emissions Technologies (NETS) at Scale!

GHG emissions (GtCO₂e/year)



10 Gt/y ~ 80 Gbbl/y storage of sCO₂

Global oil production: 34 Gbbl/y Global wastewater injection: 100 Gbbl/yr



FIGURE 8.1 Scenario of the role of negative emissions technologies in reaching net zero emissions (UNEP, 2017).

NOTES: Green represents mitigation, brown represents anthropogenic greenhouse gas emissions, and blue represents anthropogenic negative emissions. Negative emissions of 10 Gt CO_2 are required by the late 2050s and of 20 Gt CO_2 by the late 2090s because of ongoing positive emissions ties such as agriculture (mostly N₂O and CH₄) and activities that are very expensive to



ERL contributions to solution of climate crisis

- Decrease the rate of emissions
 - Geothermal
 - Electricity, low grade heat, . . .
 - Store energy and H₂
 - Carbon Capture and Storage (CCS)
- Remove CO₂ negative emissions technologies (NET)
 - Bioenergy with CCS (BECCS)
 - Grow biomass, burn, generate electricity, CCS, repeat. . . .
 - Direct Air Capture (DAC)
 - Direct Ocean Capture (DOC)
- Geographic distribution of secure storage *at scale* match to CO₂ capture sites
 - Saline aquifers
 - Mineralization

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Matching storage to CO₂ supply at scale



Figure 3.8 Theoretical CO₂ storage capacity by region







Recognizing storage options is critical for planning

- Example: Is CCS/GSC feasible for India?
 - Not much storage identified on shore
 - Thick offshore deposits, but not well characterized is it suitable?
 - Exploration, assessment of storage resources
 - Substantial basalt is it permeable?











Can permanent CCS be demonstrated?



Role of Geophysics in Carbon Capture and Sequestration 5–7 December 2022 | Saudi Arabia Abstracts submission deadline: 24 August 2022

The main challenge for CO_2 Capture is centered around what is called storage permanence. A storage is said to be permanent if it can store 99% of the injected CO_2 for 100 years. Hence, the role of geophysics is to help monitor the status of the injected CO_2 in the reservoir and its capability of storing CO_2 during and after the injection, and to manage the process of the injection itself.

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• Yet these behaviors are expected, even for secure storage





A key to conventional GCS: rapid increase in CO₂ density with depth



Density of CO₂ increases greatly at ~ 70 bar (700 m H_2O)

Behavior depends strongly on T & PLarge and variable κ and α

$$\kappa_T = \frac{1}{\rho} \left(\frac{\partial \rho}{\partial P} \right)_T$$

$$\alpha = \frac{1}{\rho} \frac{\partial \rho}{\partial T}$$

Nonlinear response





Storage in saline aquifer: important processes

Injection of cold sCO₂ into a hot aquifer (later *T* recovery, changes in *V* and *P*)

Trapping of some CO_2 in pores & throats (changing solubility as T and P change)

Ponding of sCO_2 beneath caprock (changing κ and α as T and P change)

CO₂ dissolves in brine and sinks Plume shrinkage expected!





Szulczewski et al. PNAS 2012;109:5185-5189



- ★ CO₂ plume is mobile and buoyant
- Plume migrates due to groundwater flow and aquifer slope
- Plume spreads due to buoyancy
- Plume shrinks due to capillary trapping
- Plume shrinks due to dissolution

Brine with dissolved $CO_2 \approx 3-5 \%$ denser than before dissolved





Chemo-elasticity: ΔV & Δp changes

- Dissolution
 - Easily misinterpreted as leakage
 - Large effects on in situ κ and α
- Mineral replacement large ΔV
 - Effect on permeability?
 - Pore clogging?
 - Carbofracking?
 - Seismicity?

Dissolution reactions:	Calcite, siderite, magnesite
Basaltic rock + $xH^+ \dots \rightarrow Mg^{2+} + Ca^{2+} + Fe^{2+} + AI^{3+} + SiO_{2(aq)} \dots$	
$Mg_2SiO_4 + 4H^+ \rightarrow 2Mg^{2+} + 2H_2O + SiO_{2(aq)}$	
$Fe_2SiO_4 + 4H^+ \rightarrow 2Fe^{2+} + 2H_2O + SiO_{2(aq)}$	
Precipitation reactions:	
$(Ca, Mg, Fe)^{2+} + CO_3^{2-} \rightarrow$	(Ca,Mg,Fe)CO ₃
(calcite, siderite, magnesite)	

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)man ophiolite, courtesy P. Kelema

Summary – looking forward

- CCS is essential to mitigate large temperature increases
 - Near term: Decrease rate of emission of CO₂ into the atmosphere
 - Long term: Remove CO₂ from the atmosphere net negative emissions
- The scale needed is daunting
 - > than current rate of wastewater injection and hydrocarbon production
- Sources of CO₂ will be distributed unevenly globally, storage sites must match
 - Much more exploration and characterization are needed
 - Shallow reservoirs may be best (high porosity and permeability, lower risk of leakage & sseismicity)
- It is not feasible to demonstrate 99 % containment
 - Successful storage sites are easily misidentified as leaking!
 - Tolerating minor leakage less damaging than stopping storage
 - Regulations should be based on detecting leaks, rather than proving containment